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Carbon monoxide

Carbon monoxide has a computed fractional bond order of 2.6, indicating that the "third" bond is important but constitutes somewhat less than a full bond. Thus, in valence bond terms, $-\text{C}^{\ominus}\text{O}^{\oplus}$ is the most important structure, while $:\text{C}=\text{O}$ is non-octet, but has a neutral formal charge on each atom and represents the second most important resonance contributor.